## Grade 8 FI Math \& Science Learning Opportunities <br> Week of May $4^{\text {th }}$

## MATH ENGLISH

| MANGAHIGH CHALLENGES <br> I will be changing our challenges this week to focus on ratios. Please let me know if you need your mangahigh password and I will send it to you. <br> Don't forget you can message me on mangahigh if you have any questions. Always try each challenge at least three times. Good luck! <br> Julie.roe@nbed.nb.ca | MATH AT WORK-MATH MEETS ENTREPRENEURSHIP \| FULL EPISODE <br> https://www.youtube.com/watch?v=tAfc Z3ArEs8 |
| :---: | :---: |
| NA気 C N M R 트NGE | SCRATCH 30 <br> Choose any 4 one-digit numbers: <br> Use 2, 3, or 4 of these numbers with any operations to find an answer of 1-30. <br> You cannot use a number more than once per expression. $\begin{aligned} & 123456789 \\ & 10111213141516171819 \\ & 2021222324252627282930 \end{aligned}$ |

## MAATM-SCOENCE LINR

In the past couple of weeks, you probably noticed quite a bit of Math creeping into your Science Learning Opportunities. The great news is that you will continue to see more and more of that as you keep moving ahead in your Science studies! This week, I am sharing some neat calculator tools that you may find useful \& fun. As per usual, these are the Android versions but I'm sure Apple users will find something very similar.

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## SCOENCE

## Mass, Volume and Density

1. Watch these videos to understand how to calculate density from the volume and mass of an object. Each video is important.
https://www.youtube.com/watch?v=DVQMWihs3wQ
https://www.youtube.com/watch?v=xLrOEIDc48|


## Checklist:

FORMULAS: $\quad \mathrm{D}=\mathrm{M} \div \mathrm{V}$
$M=D x V \quad V=M \div D$
DEFINITION: Density = is the mass of a substance compared to the volume of space it contains
UNITS OF MEASUREMENT:
Volume: cubic centimeters $\left(\mathrm{cm}^{3}\right)$ for solids AND milliliters ( mL ) for liquids and gasses Mass: grams (g)

Density: $\mathbf{g} \div \mathbf{c m}^{\mathbf{3}}$ for solids and $\mathbf{g} \div \mathbf{m l}$ for liquids and gasses (mass $\div$ volume)

## 2. Solve the following density problems:

1. An aluminum block has a volume of 15 mL and has a mass of 45 g . What is its density?
2. Mercury (liquid metal) is poured into a graduated cylinder which supports 20 mL and is completely filled. This mercury has a mass of 320 g . Calculate the density of mercury.
3. What is the mass of 200 mL of liquid ethanol if its density is $0.8 \mathrm{~g} / \mathrm{mL}$ ?
4. A copper block has a mass of 320 g . The dimensions of the block are 8 cm by 5 cm by 4 cm . Find the density of the block of copper. (hint: find the volume first !!!) $V=L \times W \times H$
5. What is the volume of a block of silver whose mass is 2500 g and the density is $25 \mathrm{~g} / \mathrm{cm}^{3}$.
6. Find the mass of a sample of benzene gas with a volume of 100 mL and a density of $0.9 \mathrm{~g} / \mathrm{mL}$.
7. A lead block with measurements of 4 cm by 5 cm by 6 cm has a mass of 240 g . Find the density of the lead.
